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*review (3 FRQ walkthroughs:
thermodynamics, kinetics, stoichiometry)*

Most Common Chemistry Final Exam

Question: Limiting Reactants Review

Mole Concept and Stoichiometry - 2+

Empirical and Molecular Formula | ICSE

Class 10 Chemistry IGCSE CHEMISTRY

REVISION [Syllabus 4] - Stoichiometry

Limiting Reactant How To Calculate

Theoretical Yield and Percent Yield Mole

Concept Tips and Tricks Introduction to

Oxidation Reduction (Redox) Reactions

Modern Chemistry Stoichiometry chapter 9

Review

CHAPTER 9 REVIEW Stoichiometry

MIXED REVIEW SHORT ANSWER

Answer the following questions in the space provided.

1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$

a. What is the value of the coefficient x in this equation? 40.07 g/mol

b. What is the molar mass of C_3H_4 ? 2

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mol O 2:1 mol H 2O c. What is the mole
ratio ...

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CHAPTER 9 REVIEW Stoichiometry

SECTION 3 PROBLEMS Write the

answer on the line to the left. Show all

your work in the space provided. 1. _____

The actual yield of a reaction is 22 g and
the theoretical yield is 25 g. Calculate the

percentage yield. 2. 6.0 mol of N₂ are
mixed with 12.0 mol of H

CHAPTER 9 REVIEW Stoichiometry

CHAPTER 9 Stoichiometry - Modern

Chemistry Textbook Back CHAPTER 9

Stoichiometry Stoichiometry comes from

the Greek words stoicheion, meaning

"element," and metron, meaning

"measure." Back Introduction to

Stoichiometry SECTION 9-1

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OBJECTIVES Define stoichiometry. The reaction-stoichiometry problems in this chapter can be...

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multiple choice Know the definitions of
reaction & composition stoichiometry,
mole ratio, and units of molar mass. Know
what mole ratio means and how it is used
in stoichiometry. Determine mole ratio

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using balanced formula equation.

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