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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter
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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW
SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b. What is the molar mass of C_3H_4 ? 2 mol O₂:1 mol H₂O c. What is the mole ratio ...

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Vocabulary from Chapter 6 of the Modern Chemistry
textbook. (30 total words) Terms in this set (30)
chemical bond. the attractive force that holds atoms
or ions together. ionic bonding. a force that attracts
electrons from one atom to another, which transforms
a neutral atom into an ion.

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